ALUMINIUMCHLORIDE-INDUCED ADDITIONS OF FORMALDEHYDE TO ALKENES

Jürgen O. Metzger* and Ursula Biermann

Department of Chemistry, University of Oldenburg, Carl-von-Ossietzky-Straβe 9-11, 26111 Oldenburg (Germany)

Dedicated to Professor H.G. Viehe on the occasion of his 65th birthday

ABSTRACT

The addition of formaldehyde to alkenes induced by AlCl₃ is a simple route to alkylsubstituted tetrahydropyrans. Additions of two equivalents of formaldehyde to 1-alkenes and 1,2-dialkylsubstituted alkenes give 3-alkyl-4-chlorotetrahydropyrans and 3,5-dialkyl-4-chlorotetrahydropyrans, respectively. The products are obtained as a mixture of diastereomers. 2,5-dialkyl-4-chlorotetrahydropyrans are obtained in AlCl₃-induced additions of one equivalent of formaldehyde to homoallylic alcohols. Reductive dechlorination gives the corresponding 3-, 3,5- and 2,5-alkylsubstituted tetrahydropyrans, respectively.

INTRODUCTION

The acid-catalyzed addition of aldehydes to olefines is usually called the Prins reaction (1-3). Generally the reaction gives a complex mixture of products consisting of 1,3-dioxanes, 1,3-glycols and unsaturated alcohols as main products. Cyclic ethers such as tetrahydropyrans and tetrahydrofurans were identified occasionally as minor products. It should be of interest to find reaction conditions giving with high selectivity only one of the formaldehydealkene addition products.

There are known Lewis acid induced reactions - e.g. SnCl4 and BF₃. Et₂O - giving as main products the corresponding unsaturated alcohols (4-6). Primary homoallylic alcohols are obtained in good yields and with high selectivity in alkylaluminium halide induced ene reactions (7-9). We reported recently the alkylaluminium halide induced ene addition of formaldehyde to readily available unsaturated fatty compounds (10,11), for example oleic acid and 10-undecenoic acid which are of interest as renewable raw materials (12) and obtained in good yields the corresponding primary homoallylic alcohols.

3-Alkyl-4-halotetrahydropyrans are obtained in reactions of paraformaldehyde and α -olefines in the presence of hydrogen halides (13a). Using hydrogen chloride at -60° to - 70°C the products were obtained in yields of 70-80% as mixtures of diastereomers. The condensation of *cis*- and *trans*-butene with formaldehyde and hydrogen chloride affords the diastereomeric chloroalcohols and their formals as main products while the corresponding alkylsubstituted chlorotetrahydropyrans were formed only in small amounts (14). The reaction of formaldehyde and cyclohexene in the presence of hydrogen chloride gives a mixture of two chlorinated bicyclic ethers along with a small amount of the corresponding formaldehyde acetal (15). Addition of

formaldehyde to 3-buten-1-ol catalyzed by hydrochloric acid gives 4-chlorotetrahydropyran (16). 2-Alkyl-4-chlorotetrahydropyrans are obtained in analogous reactions with aldehydes such as acetaldehyde, propional-dehyde and butyric aldehyde. The corresponding reactions catalyzed by sulphuric acid afford the substituted tetrahydropyran-4-ols as major products. Additions of aldehydes and ketones to the substituted alcohol 4-methyl-4-penten-2-ol in the presence of p-toluenesulphuric acid give 3,6-dihydro-2H-pyrans (17).

In the present paper we describe the addition of formaldehyde to olefines induced by AlCl₃ to give with high selectivity alkylsubstituted tetrahydropyrans.

RESULTS AND DISCUSSION

The AlCl₃-induced reaction of formaldehyde and 1-octene (1a) or methyl 10-undecenoate (1b) gave the corresponding 3-alkyl-4-chlorotetrahydropyrans trans-2a/cis-2a and trans-2b/cis-2b, respectively. The diastereomers were obtained in a ratio of [cis]: [trans] = 1:2.5. The isomers could be assigned by their ¹³C NMR data. The chemical shifts of the ring carbons were identical to those of cis- and trans-4chloro-3-propyltetrahydropyran (13b). The reaction takes place by addition of two equivalents of formaldehyde to the double bond of the alkene. In the first step one equivalent of formaldehyde should be added to give the ene adduct, the homoallylic alcohol. It is known that homoallylic alcohols cyclize to give 4-chlorotetrahydropyrans on reaction with formaldehyde and hydrogen chloride (18). This reaction sequence was confirmed by the following experiment: Methyl 12-hydroxydodec-9-enoate obtained by ethylaluminium dichloride-induced addition of formaldehyde to methyl 10-undecenoate (10), reacted with formaldehyde induced by AlCl₃ to give 3-alkyl-4-chlorotetrahydropyran

trans-2b/cis-2b. Reductive dechlorination of trans-2a/cis-2a and trans-2b/cis-2b for example with Bu₃SnH afforded the 3-alkyl-tetrahydropyrans 3a and 3b, respectively.

column chromatography and obtained as $(3\alpha, 4\beta, 5\beta)$ -7a/ $(3\alpha, 4\beta, 5\beta)$ -8a and $(3\alpha, 4\alpha, 5\alpha)$ -7b/ $(3\alpha, 4\alpha, 5\alpha)$ -8b in a ratio of 3:1.

$$(CH2O)ff/CH2Cl2, ACl3, r.t., 24 h$$

$$(3\alpha, 4\beta, 5\beta)-5 \qquad (3\alpha, 4\alpha, 5\alpha)-5$$

The addition of formaldehyde to 1,2-dialkylsubstituted alkenes 4, 6a and 6b gave the corresponding 3,5-dialkyl-4-chlorotetrahydropyrans 5, 7a/8a and 7b/8b in good yields. The reaction of cis-2-heptene (4) afforded regioselectively 4-chloro-5-methyl-3-propyltetrahydropyran (5) as a diastereomeric mixture of $(3\alpha, 4\beta, 4\beta)$ -5 and $(3\alpha, 4\alpha, 5\alpha)$ -5 in a ratio of 4:1. The addition products with methyl oleate (6a) and methyl petroselinate (6b) were obtained as regioisomers (1:1 mixture) and as diastereomers. Reductive dechlorination of 7a/8a and 7b/8b with Bu₃SnH gave the corresponding 3,5-dialkyltetrahydropyrans 9a/10a and 9b/10b.

The diastereomeric mixture of 7a/8a was separated by

The addition of formaldehyde to methyl ricinoleate (11), an alkene with β -hydroxy functionality took place regioselectively to position C-9 of the olefine. Cyclization and elimination of H₂O afforded 2,5-dialkyl-4-chlorotetrahydropyran 12 as a mixture of diastereomers. (2 α , 4 α , 5 β)-12 was obtained by recrystallization of 12 from petrolether. Reductive dechlorination of 12 gave a mixture of the 2,5-dialkyltetrahydropyrans *trans*-13 and *cis*-13 in a ratio of 3.8:1.

The stereochemistry of the main products can be explained by anti-attack of the formaldehyde-AlCl₃-complex to the homoallylic alcohol existent in the preferred conformation avoiding allylic strain (19). The minor products are formed by syn-attack.

$$R = (CH_{2})_{7}COOCH_{3}$$

$$R^{1} = (CH_{2})_{7}COOCH_{3}$$

$$R^{2} = (CH_{2})_{7}COOCH_{3}$$

$$R^{3} = (CH_{2})_{7}COOCH_{3}$$

$$R^{2} = (CH_{2})_{7}COOCH_{3}$$

$$R^{3} = (CH_{2})_{7}COOCH_{3}$$

$$R^{4} = (CH_{2})_{7}COOCH_{3}$$

$$R^{5} = (CH_{2})_{7}COOCH_{3}$$

$$R^{5} = (CH_{2})_{7}COOCH_{3}$$

The AlCl₃-induced addition to formaldehyde to alkenes gives with high selectivity the corresponding alkylsubstituted 4-chlorotetrahydropyrans. In this study the synthesis of 3-, 3,5- and 2,5-alkylsubstituted tetrahydropyrans is described. By variation of the alkene on the one hand and of the carbonyl compound on the other hand it should be possible to synthesize the complete set of alkylsubstituted tetrahydropyrans.

EXPERIMENTAL

Melting points (uncorrected): Leitz Laborlux 12. - Refractive indices np: Zeiss-Abbé-Refraktometer. - 1 H and 13 C NMR: Bruker AM 300 (2a, 2b, 3a, 3b, 9a/10a and 9b/10b), Bruker AMX R 500 ((3 α , 4 β , 5 β)-7a/(3 α , 4 β , 5 β)-8a, (3 α , 4 α , 5 α)-7a/(3 α , 4 α , 5 α)-8a, 7b/8b, (2 α , 4 α , 5 β)-12 and 13). TMS as internal standard; selected data are given. Full 1 H-and 13 C-NMR data are available from the authors on request. The signals of the regioisomers 7b and 8b were distinguishable in the 1 H- and 13 C-NMR spectra but they could not be assigned unambiguously to the respective products. NMR-data of the minor products cis-2a, cis-2b, (3 α , 4 α , 5 α)-5 and (3 α , 4 α , 5 α)-7b/(3 α , 4 α , 5 α)-8b are signed by a . - Analytical GC: Carlo Erba GC 6000 Vega Series 2 with a FID (DB1 - column, 28 m). - Mass spectra: Finnegan MAT 212 mass spectrometer. -

Methyl oleate (new sun flower, 82.8% methyl oleate, 3.6% methyl stearate, 3.5% methyl palmitate, 8.4% C_{18:2}), petroselinic acid (81.3% petroselinic acid, 3.3% palmitic acid, 0.4% stearic acid, 13.5% C_{18:2}) and ricinoleic acid (80 - 85% purity) were obtained from Henkel KGaA. - The amounts of the starting olefines used in the reactions were calculated based on 100% purity. The free fatty acids were converted to their methyl esters by usual esterification. - Methyl 10-undecenoate (Atochem), cis-2-heptene and 1-octene (Merck), paraformaldehyde (Janssen), Bu₃SnH, AIBN, AICl₃, KF (Fluka), 18-crown-6 (Aldrich) were used without further purification. - All reactions were run under N₂.

Synthesis of alkylsubstituted chloro-tetrahydropyrans; general procedure :

A mixture of the appropriate alkene 1a or 4 or of the methyl alkenoate 1b, 6a, 6b or 11 (5 mmol) and paraformaldehyde

(0.3 g, 10 mmol; 0.21 g, 7 mmol was used for 11) in CH₂Cl₂ (10 mL) was stirred magnetically in N₂ atmosphere for 5 min at - 15°C. After addition of AlCl₃ (0.33 g, 2.5 mmol) the sample was stirred for additional 24h at r.t. The reaction was quenched by addition for Et₂O (100 mL) and H₂O (40 mL). 10% HCl was added until the precipitated aluminium saits dissolved. The organic layer was separated and the aqueous layer extracted with Et₂O (3x40 mL). The combined organic layers were dried (Na₂SO₄) and evaporated. The product was purified by "Kugelrohr" distillation (2b, 5, 7a/8a, 7b/8b) or by column chromatography (28 cm x 2 cm) on silica gel (Merck, 70-230 mesh) with petrolether/ EtOAc (95:5) as eluent (2a, 12). Fractions containing the tetrahydropyranderivative were collected, the solvent evaporated and the residue dried at 20°C/0.01 mbar. -Reductive dechlorination of the alkylsubstituted 4-chlorotetrahydropyrans with Bu₃SnH: A solution of the tetrahydropyran 2a, 2b, 7a/8a, 7b/8b and 12 (1-2 mmol), Bu₃SnH (1 mL, 4 mmol) and AIBN (15 mg) in benzene (30 mL) was heated at 80°C for 20h. The solvent was evaporated and the residue dissolved in Et₂O (10 mL). KF (10% in H₂O, 10 mL) and 18-Crown-6 were added. The organic layer was separated and the precipitate washed with Et₂O (80 mL). The combined organic layers were dried (Na₂SO₄) and evaporated.

trans-4-Chloro-3-pentyltetrahydropyran (trans-2a) and cis -4-Chloro-3-pentyltetrahydropyran (cis-2a) (2:1 mixture): yield 0.71 g (75%), colorless liquid, np²² = 1.4635. - ¹H NMR (CDCl₃): δ = 4.42 (m, 1H, 4-H)^a, 4.02 (dd, J = 11.8, 4.9 Hz, 1H, 2-H), 3.94 (m, 1H, 6-H), 3.78 (m, 1H, 4-H), 3.78 (m, 2H, 6-H, 6-H')^a, 3.59 (dd, J = 11.5, 4.4 Hz, 1H, 2-H)^a, 3.51 (dd, J = 11.5, 9.9 Hz, 1H, 2-H')^a, 3.41 (ddd, J = 11.5, 11.0, 2.6 Hz, 1H, 6-H'), 3.11 (dd, J = 11.8, 9.4 Hz, 1H, 2-H'), 2.16 (m, 1H, 3-H)^a, 2.11 (m, 1H, 3-H), 1.94 (m, 2H, 5-H), 1.94 (m, 2H, 5-H)^a. - ¹³C NMR (CDCl₃): δ = 70.9 (C-2), 67.1 (C-2)^a, 66.9 (C-6), 62.5 (C-6)^a, 61.9 (C-4), 60.8 (C-4)^a, 44.9 (C-3), 40.0 (C-3)^a, 36.5 (C-5), 34.5 (C-5)^a. - MS/CI (isobutane): m/z (%) = 191(100)/193(32) [MH⁺], 155(56) [MH⁺-HCl]. - C₁₀H₂₀OCI: calcd. 191.1224, found 191.1323 (MS/CI). -

3-Pentyltetrahydropyran (3a): Dechlorination of 2a (0.45 g) gave 0.35 g (95%) of 3a, colorless liquid, $n_D^{22} = 1.4591$. - H NMR (CDCl₃): $\delta = 3.85$ (m, 2H, 2-H and 6-H), 3.33 (ddd, J = 11.2, 9.7, 4.4 Hz, 1H, 6-H'), 3.02 (dd, J = 11.0, 10.2 Hz, 1H, 2-H'), 1.84 (m, 1H, 3-H), 1.56 (m, 4H, 4-H and 5-H).

 $^{13}\text{C NMR (CDCl}_3): \delta = 73.6 \text{ (C-2)}, 68.5 \text{ (C-6)}, 36.0 \text{ (C-3)}, 32.5 \text{ (C-4)}, 30.1 \text{ (C-5)}. - MS/CI (isobutane)}: m/z (%) = 157(100) [MH <math display="inline">^+$]. - C10H26O : calcd. 156.1514, found 156.1501 (MS/EI).

trans-4-Chloro-3-(7-methoxycarbonylheptyl)-tetrahydropyran (trans-2b) and cis-4-Chloro-3-(7-methoxycarbonylheptyl)-tetrahydropyran (cis-2b) (2.4:1 mixture) : yield 1.01 g (73%), colorless liquid, $np^{22} = 1.4701$. - ¹H NMR (CDCl₃) : $\delta = 4.20$ (m, 1H, 4-H)^a, 3.94 (m, 2H, 2-H, 6-H), 3.75 (m, 1H, 4-H), 3.75 (m, 2H, 6-H, 6-H')^a, 3.52 (m, 2H, 2-H, 2-H')^a, 3.38 (ddd, J = 11.5, 11.3, 2.5 Hz, 1H, 6-H'), 3.08 (dd, J = 11.6, 11.6 Hz, 1H, 2-H'), 2.11 (m, 1H, 3-H)^a, 2.07 (m, 1H, 3-H), 1.91 (m, 2H, 5-H). - ¹³C NMR (CDCl₃) : $\delta = 174.0$ (C = O), 70.8 (C-2), 67.0 (C-2)^a, 66.8 (C-6), 62.4 (C-6)^a, 61.8 (C-4), 60.7 (C-4)^a, 44.8 (C-3), 40.9 (C-3)^a, 36.4 (C-5), 34.5 (C-5)^a. - MS/Cl (isobutane) : m/z (%) = 277(100)/279(34) [MH + 1], 241(56) [MH + - HCl]. - C₁₄H₂₆O₃Cl: calcd. 277.1591, found 277.1581 (MS/Cl). -

3-(7-Methoxycarbonylheptyl)-tetrahydropyran (3b) : Dechlorination of 2b (0.5 g) gave 0.43 g (97%) of 3b, colorless liquid, $n_D^{22}=1.4572.$ - ^1H NMR (CDCl3) : $\delta=3.83$ (m, 2H, 2-H and 6-H), 3.31 (ddd, J=10.8, 10.0, 4.3 Hz, 1H, 6-H'), 2.99 (dd, J=10.6, 10.6 Hz, 1H, 2-H'), 2.28 (t, J=7.5 Hz, 2H, CH2COOCH3), 1.82 (m, 1H, 3-H), 1.56 (m, 6H, 4-H, 5-H, CH2CH2COOCH3). - ^{13}C NMR (CDCl3) : $\delta=174.0$ (C=O), 73.5 (C-2), 68.4 (C-6), 51.3 (OCH3), 36.0 (C-3), 32.4 (C-4), 30.1 (C-5). - MS/Cl (isobutane) : m/z (%) = 243(100) [MH $^+$]. - C14H26O3 : calcd. 242.1882, found 242.1881 (MS/El).

4-Chloro-5-methyl-3-propyltetrahydropyran ((3 α , 4 β , 5 β)-5 and (3 α , 4 α , 5 α)-5) (4:1 mixture) : yield 0.53 g (60%), colorless liquid, np²² = 1.4662. - ¹H NMR (CDCl₃) : δ = 4.08 (dd, J=5.8, 3.7 Hz, 1H, 4-H)^a, 4.01 (ddd, J=11.5, 4.5, 0.9 Hz, 1H, 2-H), 3.97 (dd, J=11.6, 3.4 Hz, 1H, 4-H, 4-H), 3.88 (ddd, J=11.5, 4.5, 1.0 Hz, 1H, 6-H), 3.55 (m, 2H, 2-H, 6-H)^a, 3.32 (dd, J=10.6, 10.5 Hz, 2H, 2-H', 6-H')^a, 3.04 (dd, J=11.5, 11.5 Hz, 1H, 6-H'), 3.00 (dd, J=11.5, 11.4 Hz, 1H, 2-H'). - ¹³C NMR (CDCl₃) : δ = 73.6 (C-2), 72.2 (C-6), 71.0 (C-2)^a, 67.4 (C-6)^a, 70.3 (C-4), 66.2 (C-4)^a, 44.9 (C-6), 41.7 (C-3)^a, 41.4 (C-5). -

4-Chloro-3-(6-methoxycarbonylhexyl)-5-octyltetrahydropyran (7a) and 4-Chloro-5-heptyl-3-(7-methoxycarbonylheptyl)-tetrahydropyran (8a) (1:1 mixture) : yield 1.61 g (86%), colorless liquid, $n_D^{22} = 1.4723$. - MS/Cl (isobutane) : m/z (%) = 375(100)/377(34) [MH⁺], 339(66) [MH⁺ - HCl] - C21H39O3Cl (374.2): calcd. C 67.39, H 10.42, found C 67.02, H 11.03. - The diastereomeric mixture of the regioisomers 7a/8a (1:1) was separated by column chromatography on silica gel 60 (Merck, 70-230 mesh) with petrolether/ ether (9:1) and gave $(3\alpha, 4\beta, 5\beta)$ -7a/ $(3\alpha, 4\beta, 5\beta)$ -8a and $(3\alpha, 4\beta, 5\beta)$ -8b and $(3\alpha, 4\beta, 5\beta)$ -8c and $(3\alpha$ 4α , 5α)-7a/(3α , 4α , 5α)-8a (ratio 3:1). - (3α , 4β , 5β)-7a/(3α , 4β , 5β)-8a: ¹H NMR (CDCl₃): δ = 3.99 (dd, J = 11.6, 3.8 Hz, 1H, 2-H), 3.94 (dd, J=11.6, 2.8 Hz, 1H, 4-H), 3.53 (m, 1H, 6-H), 3.39 (dd, J = 10.8, 10.2 Hz, 1H, 6-H'), 3.0 (dd, J = 11.6, 11.5 Hz, 1H, 2-H'), 1.83 (m, 2H, 3-H, 5-H). - 13 C NMR $(CDCl_3): \delta = 174.2 (C = O), 72.2 (C-2, C-6), 69.1 (C-4), 51.4$ (O-CH₃), 45.2 (C-3, C-5). - $(3\alpha, 4\alpha, 5\alpha)$ -7a/ $(3\alpha, 4\alpha, 5\alpha)$ -8a: ¹H NMR (CDCl₃) : δ = 3.90 (m, 1H, 4-H), 3.59 and 3.58 (2xdd, J = 11.5, 3.5 Hz, 2H, 2-H) and 6-H), 3.45 (dd, $J = 11.5, 10.8 Hz, 2H, 2-H', 6-H'), 1.90 (m, 2H, 3-H, 5-H). - <math>^{13}C$ NMR

(CDCl₃) : δ = 174.2 (C = O), 66.7 (C-2, C-6), 65.6 (C-4), 51.4 (O-CH₃), 41.8 (C-3, C-5). -

3-(6-Methoxycarbonylhexyl)-5-octyltetrahydropyran (trans-9a and cis-9a) and 5-Heptyl-3-(7-methoxycarbonylheptyl)-tetrahydropyran (trans-10a and cis-10a) (1:1 mixture): Dechlorination of **7a/8a** (1:1 mixture, 0.5 g) gave 0.42 g (93%) of 9a/10a, colorless liquid, $n_D^{22} = 1.4602.$ ¹H NMR (CDCl₃): δ = 3.82 (m, 2H, 2-H, 6-H, trans-9a/10a), 3.27 (m, 4H, 2-H, 2-H', 6-H, trans-9a/10a), 2.79 (dd, J = 11.1, 11.0 Hz, 2H, 2-H', 6-H', trans-9a/10a), 2.23 (t, J = 7.5 Hz, 2H, CH₂COOCH₃), 1.55 (m, 4H, 3-H, 5-H, CH₂-CH₂COOCH₃). ¹³C NMR (CDCl₃): δ = 174.1 (C = 0), 73.6 (C-2, C-6, trans-9a/10a), 72.8 (C-2, C-6, cis-9a/10a), 51.4 (O-CH₃), 45.6 (C-4, cis-9a/10a), 37.4 (C-4, trans-9a/10a), 36.1 (C-3, C-5). - C₂₁H₄₀O₃: calcd. 340.2977, found 340.2980 (MS/EI). -

4-Chloro-3-(3-methoxycarbonylpropyl)-5-undecyltetrahydropyran (**7b**) and 4-Chloro-5-decyl-3-(4-methoxycarbonylbutyl)-tetrahydropyran (**8b**) (1:1 mixture): yield 0.91 g (58%), colorless liquid, np²² = 1.4730. - The main product is the 1:1 mixture of (3α, 4β, 5β)-**7b** and (3α, 4β, 5β)-**8b**. - ¹H NMR (CDCl₃): δ = 3.97 and 3.96 (dd, J=11.4, 5.6 Hz, 1H, 2-H, **7b** and **8b**), 3.92 (m, 1H, 4-H), 3.57 (m, 2H, 2-H, 6-H)^a, 3.50 (m, 2H, 2-H', 6-H')^a, 3.42 (dd, J=11.1, 11.1 Hz, 1H, 6-H), 3.36 (dd, J=11.1, 11.1 Hz, 1H, 6-H'), 2.99 (dd, J=11.4, 11.5 Hz, 1H, 2-H'), 2.28 and 2.27 (t, J=7.4 Hz, 2H, CH₂COOCH₃), 1.78 (m, 2H, 3-H, 5-H). - ¹³C NMR (CDCl₃): δ = 173.8 (C=0), 72.1 (C-2, C-6), 68.9 (C-4), 66.6 (C-2, C-6)^a, 65.3 (C-4)^a, 51.3 (O-CH₃), 45.1 and 45.0 (C-3, C-5, **7b** and **8b**), 41.8 and 41.6 (C-3, C-5, **7b** and **8b**). - MS/Cl (isobutane): m/z (%) = 375(100)/377(34) [MH⁺], 339(100) [MH⁺ - HCl). - C₂₁H₃₉O₃Cl (374.2): calcd. C 67.39, H 10.42, found C 67.47, H, 10.23. -

3-(3-Methoxycarbonylpropyl)-5-undecyltetrahydropyran (trans-9b and cis-9b) and 5-Decyl-3-(4-methoxycarbonylbutyl)-tetrahydropyran (trans-10b and cis-10b) (1:1 mixture): Dechlorination of **7b/8b** (1:1 mixture, 0.5 g) gave 0.4 g (89%) of **9b/10**b, colorless liquid, $n_D^{22} = 1.4618.$ - 1 H NMR (CDCl₃): $\delta = 3.90$ (m, 2H, 2-H, 6-H, trans-9b/10b), 3.39 (m, 4H, 2-H, 2-H', 6-H, 6-H', cis-9b/10b), 2.87 (dd, J = 11.0, 11.0 Hz, 2H, 2-H', 6-H', trans-9b/10b), 2.31 and 2.30 (t, J = 7.6 Hz, 2H, CH₂COOCH₃), 9b and 10b). 1.62 (m, 4H, 3-H, 5-H, CH₂CH₂COOCH₃). - 13 C NMR (CDCl₃): $\delta = 73.5$ (C-2, C-6), 51.3 (O-CH₃), 45.5 (C-4, cis-9b/10b), 37.2 (C-4, trans-9b/10b), 36.0 (C-3, C-5). - C₂₁H₄OO₃: calcd. 340.2977, found 340.2977 (MS/EI). -

4-Chloro-2-hexyl-5-(7-methoxycarbonylheptyl)-tetrahydropyran ((2α, 4α, 5β)-12 and (2α, 4α, 5α)-12) (3.8:1 mixture): yield 0.88 g (61%). Recrystallization of (12) from petrolether gave pure (2α, 4α, 5β)-12, solid, m.p. 29-30°C. - 1 H NMR (CDCl₃): δ = 4.02 (dd, J = 11.5, 4.6 Hz, 1H, 6-H), 3.71 (ddd, J = 11.4, 11.2, 4.5 Hz, 1H, 4-H), 3.24 (m, 1H, 2-H), 3.05 (dd, J = 11.5, 11.3 Hz, 1H, 6-H'), 2.30 (t, J = 7.5 Hz, 2H, CH₂COOCH₃), 2.16 (ddd, J = 12.8, 4.5, 1.5 Hz, 1H, 3-H), 1.85-1.49 (m, 5H, 3-H', 5-H, CH₂CH₂COOCH₃, HCHCHO). - 13 C NMR (CDCl₃): δ = 174.2 (C = 0), 77.8 (C-2), 71.5 (C-6), 62.7 (C-4), 45.1 (C-5), 43.1 (C-3). - MS/Cl (isobutane): m/z (%) = 361(100)/363(34) [MH $^+$], 325(66) [MH $^+$ - HCl]. - C₂₀H₃₈O₃Cl: calcd. 361.2531, found 361.2505 (MS/Cl). -

2-Hexyl-5-(7-methoxycarbonylheptyl)-tetrahydropyran (trans-13 and cis-13): Dechlorination of 12 (0.5 g) gave 0.41

g (91%) of **13**, solid, m.p. 34-35°C: ¹H NMR (CDCl₃): δ = 3.89 (m, 1H, 6-H, trans-**13**), 3.25 (m, 1H, 2-H, cis-**13**), 3.15 (m, 1H, 2-H, trans-**13**) 2.98 (dd, J = 11.1, 11.1 Hz, 1H, 6-H', trans-**13**), 1.84 (m, 1H, 3-H trans-**13**).

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REFERENCES

- 1. Adams, D.R.; Bhatnagar, S.P. Synthesis 1977, 661.
- 2. Arundale, E.; Mikeska, L.A. Chem. Rev. 1952, 51, 505.
- 3. Snider, B.B. in *Comprehensive Organic Synthesis* (Ed. : B.M. Trost, J. Fleming), Pergamon Press, Exeter, **1991**, vol. 2, p. 527.
- Blomquist, A.T.; Himics, R.J. Tetrahedron Lett. 1967, 40, 3947.
- Yang, N.C.; Yang, D.-D.H.; Ross, C.B. J. Am. Chem. Soc. 1959, 81, 133.
- Blomquist, A.T.; Himics, R.J. J. Org. Chem. 1968, 33, 1156.
- 7. Snider, B.B. Acc. Chem. Res. 1980, 13, 426.
- Snider, B.B.; Rodini, D.J.; Kirk, T.C.; Cordova, R.J. J. Am. Chem. Soc. 1982, 104, 555.
- 9. Snider, B.B.; Phillips, G.B. J. Org. Chem. 1983, 48, 464.
- 10. Metzger, J.O.; Biermann, U. Synthesis 1992, 463.
- Biermann, U.; Metzger, J.O. Fat Sci. Technol. 1991, 93, 282.
- Baumann, H.; Bühler, M.; Fochem, H.; Hirsinger, F.;
 Zoebelein, H.; Falbe, J. Angew. Chem. 1988, 100, 41;
 Angew. Chem., Int. Ed. Engl. 1988, 27, 41.
- a) Stapp, P.R. J. Org. Chem. 1969, 34, 479. b) Stapp, P. in Sadtler Standard Carbon-13 NMR Spectra (Ed. Sadtler Research Laboratories, INC.), Philadelphia, 1978, vol. 18, 3578C.
- Stapp, P.R.; Weinberg, D.S. J. Org. Chem. 1969, 34, 3592.
- 15. Stapp, P.R.; Randall, J.C. J. Org. Chem. 1970, 35, 2948.
- 16. Hanschke, E. Chem. Ber. 1955, 88, 1053.
- 17. Williams, P.H.; Ecke, G.G.; Ballard, S.A. *J. Am. Chem. Soc.* **1950**, *72*, 5738.
- 18. Colonge, J. C.R. Acad. Sci. 1955, 240, 1552.
- 19. Hoffmann, R.W. Chem. Rev. 1989, 89, 1841.